I B.Tech Supplimentary Examinations, Aug/Sep 2007 ENGINEERING CHEMISTRY (Common to Mechanical Engineering, Mechatronics, Production Engineering and Automobile Engineering)

Time: 3 hours

Max Marks: 80

Set No. 1

Answer any FIVE Questions All Questions carry equal marks

1. Write a brief account on the following: (a) Treatment of water for drinking purpose. (b) Determination of chlorides in water. [8+8](a) What is meant by desalination? What is its significance? 2. (b) Explain the different methods used for the desalination of brackish water. [4+12]3. (a) What are electrochemical series? How are they useful in determining the rate of corrosion? (b) How do you differentiate electrochemical series from galvanic series? [8+8]4. (a) What is Sheradizing? Explain and discuss its applications. [8+8](b) Write note colirizing and chromizing. (a) Write a note on properties and uses of Teflon. 5.(b) Differentiate the Natural Polymer and synthetic polymer. [4+6+6](c) Write a note on silicone rubbers. 6. Explain the following two theories for the mechanism of the lubricants. (a) Boundary lubrication (b) Extreme pressure lubrication. [8+8](a) Discuss solid lubricants. 7. (b) Explain the mechanism of thin-film lubrication. [8+8](a) Define a Fuel? How chemical fuels are classified and give examples for each. 8. (b) What is meant by Calorific value of a fuel? Define calorie and kilocalorie. [10+6]

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Time: 3 hours

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Set No. 2

Answer any FIVE Questions All Questions carry equal marks

- (a) What is hardness of water? How do you express the hardness? What are the units to express the hardness?
 - (b) Give an account of the disadvantages of hard water. [8+8]
- 2. Compare the following processes.
 - (a) Permutit process with lime soda process.
 - (b) Phosphate conditioning with calgon conditioning. [16]
- 3. Explain the process of wet corrosion by evolution of hydrogen and absorption of oxygen. [16]

4. Differentiate the following with suitable examples. [16]

- (a) Paints from varnishes
- (b) Drying oils from non drying oils.
- 5. (a) Write the repeat unit in Bakelite, PVC and Nylon?
 - (b) Mention any two important compounding ingredients of rubber.
 - (c) Write a note on Silicones. [5+6+5]
- 6. Write a note on lubricants with special reference to their classification, mode of action, examples and applications. [16]
- 7. Describe the various types of lubrication. [16]
- 8. (a) Define a Fuel? How chemical fuels are classified and give examples for each.
 - (b) What is meant by Calorific value of a fuel? Define calorie and kilocalorie. [10+6]

Set No. 3

I B.Tech Supplimentary Examinations, Aug/Sep 2007 ENGINEERING CHEMISTRY (Common to Mechanical Engineering, Mechatronics, Production Engineering and Automobile Engineering)

Time: 3 hours

Max Marks: 80

Answer any FIVE Questions All Questions carry equal marks ****

- 1. (a) What is the cause of hardness of water? How is the hardness of water expressed?
 - (b) Explain the soap titration method for the estimation of total hardness of water.
 - (c) Calculate the temporary hardness of water from the following data by the soap titration method, when 100 ml of the water sample is titrated with soap solution.
 Leather factor = 0.6 ml soap solution
 Total hardness = 18.6 ml soap solution
 Permanent hardness = 6 ml of soap solution.
 Standard hardwater (400 mg/l of CaCO₃) = 36 ml. [4+6+6]
- 2. (a) What is meant by desalination? What is its significance?
 - (b) Explain the different methods used for the desalination of brackish water.

[4+12]

[16]

[16]

- 3. Justify the following statements by giving suitable examples.
 - (a) Electrochemical series gives a basis for the prediction of the process of corrosion.
 - (b) Design and material selection help to control metallic corrosion. [8+8]
- 4. Explain the following terms:
 - (a) Drying oil
 - (b) Thinners
 - (c) Driers
 - (d) Fillers.
- 5. (a) Discuss the compounding of plastics?
 - (b) Explain the procedures used in the processing of Natural rubber.
 - (c) Write a note on urea formaldehyde resins. [5+5+6]
- 6. (a) Define flash and fire points.
 - (b) Discuss the important functions of lubricants. [16]
- 7. How to select lubricants for the following:

Set No. 3

- (a) cutting tools
- (b) I.C engines
- (c) steam engines
- (d) steam turbines
- (e) Gears.
- 8. (a) Define a Fuel? How chemical fuels are classified and give examples for each.
 - (b) What is meant by Calorific value of a fuel? Define calorie and kilocalorie.

[10+6]

I B.Tech Supplimentary Examinations, Aug/Sep 2007 ENGINEERING CHEMISTRY (Common to Mechanical Engineering, Mechatronics, Production Engineering and Automobile Engineering)

Time: 3 hours

Max Marks: 80

[16]

[16]

Set No. 4

Answer any FIVE Questions All Questions carry equal marks

- 1. (a) What is meant by break point chlorination? What is its significance and advantages?
 - (b) Explain the sedimentation process for the treatment of muncipal water. [8+8]
- 2. Write short notes on the following:
 - (a) Carry over
 - (b) Ion-exchange process.
- 3. (a) What is corrosion? What are the units in which it is expressed? How is it different from errosion?
 - (b) Explain the mechanism of electrochemical corrosion. [8+8]
- 4. Explain different types of Metallic Coatings. [16]
- 5. (a) How is PVC prepared and what are its uses?
 - (b) Write down the differences between compression and injection moulding techniques.
 - (c) How is Thiokol prepared? Mention its uses. [4+6+6]
- 6. Write a note on lubricants with special reference to their classification, mode of action, examples and applications. [16]

7. How to select lubricants for the following:

- (a) cutting tools
- (b) I.C engines
- (c) steam engines
- (d) steam turbines
- (e) Gears.
- 8. (a) Explain the recovery of by-product from 'Coke oven gas'
 - (b) Give the comparison between solid, liquid and gaseous fuels. [8+8]

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